

## SHAPE MEMORY-ASSISTED SELF-HEALING (SMASH) OF THIOL-ENE NETWORKS

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Shape memory-assisted self-healing is a promising concept to improve the self-healing performance of vitrimeric networks. With the help of the entropic recoil coming from the shape memory functionality of thiol-ene vitrimers, we achieved high healing efficiency, following a close-then-heal strategy for a scratch of a size in the range of several hundred micrometers. The networks were prepared out of a trifunctional thiol crosslinker, functional acrylates bearing aromatic rings for stiffness and -OH groups for transesterification, an organic phosphonate as catalyst and a photo-initiator for photocuring. Organic phosphonates were used as transesterification catalysts, which proved to show catalytic effects on transesterification of thiol-ene networks in another study by our group [1]. Herein, we continue the study by varying the molar content of the thiol compound to produce several networks with varying glass transition temperatures ( $T_g$ ) and stiffness. The performance of the selected networks was examined by carrying out FTIR, DMA, stress-relaxation tests, TGA and tensile testing for determining the healing efficiency. The results showed that  $T_g$  and storage modulus ( $E'$ ) in the rubbery region increase with decreasing thiol content. Based on the network structure and mobility, we expect a difference in the efficiency of the scratch closure induced by shape recovery (affected by rigidity) and healing efficiency (influenced mostly by network mobility and availability of -OH and ester moieties). Stress relaxation indicates a topology freezing transition temperature ( $T_v$ ) between 74 °C and 137 °C. The general trend is that a decreased thiol content produces stiffer networks with higher  $T_g$  that are slower in shape recovery and take longer time for relaxing stresses, therefore needing an extended thermal treatment to successfully induce healing.

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[1] Rossegger, Elisabeth, et al. "Digital light processing 3D printing with thiol-acrylate vitrimers." *Polymer Chemistry* (2020).